

## Effect Of pH On Liquid-Phase Mass Transfer And Diffusivity Coefficient At Leachate Treatment Of Municipal Waste Landfill In Anaerobic Bioreactor

Abdul Kahar<sup>a\*</sup>, IDAA Warmadewanthi<sup>b</sup> and Joni Hermana<sup>b</sup>

<sup>a</sup>Department of Chemical Engineering, Faculty of Engineering, Mulawarman University, Gunung Kelua Jl. Sambaliung No. 9, Samarinda 75119, Indonesia

<sup>b</sup>Department of Environmental Engineering, Faculty of Civil Engineering and Planning, Institut Teknologi Sepuluh Nopember, ITS Jl. AR Hakim Sukolilo, Surabaya 60111, Indonesia

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**ABSTRAK:** Leachate is a liquid waste resulted from physical, biological and chemical decomposition of landfill waste. Leachate contains complex dissolved organic and inorganic substrate which are biodegradable and non-biodegradable. Anaerobic treatment principally utilizes anaerobic bacteria in order to degrade the dissolved organic substance into biogas. Anaerobic treatment is very sensitive towards the substrate concentration, temperature, and pH. This research used anaerobic bioreactor with the volume of 160 L, the ratio of the leachate:biogas is 70:30. Seeding and acclimatization steps were done, respectively for 10 days, leachate treatment was done in 21 days. Seeding, acclimatization, and leachate treatment were done on the pH ambience of 7.2 and 8.0 and ambient temperature. COD and VFA analysis were done every two days. The objective of this research is to decide the pH effect on the coefficient of liquid-phase mass transfer:  $k_L$ , and the diffusivity of the dissolved substance,  $D_L$ . pH affects the degradation of the concentration of dissolved organic substrate in the leachate. The higher the pH is, the higher the obtained VFA concentration is. VFA concentration is affected by pH; however, it still considers the optimal pH condition of the substrate biodegradation. pH affects the average rate of mass transfer,  $r_{kL}$  and diffusivity of the dissolved substance,  $D_L$  in the anaerobic treatment of leachate. pH affect the concentration of dissolved organic substrate which subsequently influences the coefficient of liquid-phase mass transfer of the leachate,  $k_L$  and the diffusivity of the dissolved substance,  $D_L$ . The higher the pH is, the more increasing the  $k_L$  is and the more decreasing the  $D_L$  is.

**Keywords:** anaerobic bioreactor; diffusivity leachate; mass transfer

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## 1. Introduction

### 1.1. Leachate

Leachate is a liquid waste resulted from the influx of external water into the landfill waste. It either dissolves and washes away dissolved organic and inorganic materials or is suspended within the waste, including complex organic material resulted from physical, biological, and chemical decomposition (Kumaret al., 2013). Therefore, leachate becomes a complex mixture from dissolved organic material and inorganic contaminants. Leachate contains: VFA, LCFA, fulvic and humic compounds, ammonia-nitrogen, phosphate, sulfate, heavy metal, organic xenobiotic (XOCs); aromatic hydrocarbons, phenols, and chlorinated aliphatics, inorganic salts and microorganism (Christensen et al., 2001; Renou et al., 2008; Zainol et al., 2012; Hassan and Xie, 2014); and bioreactory contaminants (Tatsi et al., 2003). Therefore, leachate contains complex dissolved organic and inorganic substrates which are biodegradable and non-biodegradable (Christensen et al., 2001).

Such characteristics make leachate very dangerous for the environment with the contamination potential transcends some industrial waste (Zainol et al., 2012; Hassan and Xie, 2014). Characteristics and quality of leachate are influenced by: characteristics and composition of waste, type of soil covering the landfill, season, pH and humid; and the age of landfill (Zainol et al., 2012; Hassan and Xie, 2014). The variation of leachate composition depends on several factors, such as composition and age of landfill, design and operation and the condition of landfill, climate and hydrogeological condition, humidity, temperature, and level of stabilization (Renou et al., 2008). The mass transfer may occur during liquid phase, gas phase, or to both phases simultaneously. It may occur during multi-phase system or one phase system (Thibodeaux, 1996; Geankoplis, 2003). With concentration gradient, there will be a movement of mass transfer from higher concentrated area to the lower concentrated area. This movement is used so that the intensive contact occurs between the dissolved material and microorganism, that it enables more mass transfer (Geankoplis, 2003; Welty et al., 2007). In the case of

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\*Corresponding Author:  
Email: a.kahar@ft.unmul.ac.id@mail.com

complex organic substrate, which is generally expressed as Biological Oxygen Demand (BOD); Volatile Fatty Acid (VFA) or Chemical Oxygen Demand (COD), a part of substrate which is difficult to degrade (Zaiat et al., 2000; Christensen et al., 2001; Hassan and Xie, 2014).

The essential problem of the technical scale in the development of bioreactor is hydrodynamics, the phenomenon of mass transfer between phases, kinetics, and the thermal transfer (Leib et al., 2001). Despite the fact that it is not the main factor on bioprocess, mass transfer is crucial in anaerobic bioreactor (Leib et al., 2001; Doble, 2006; Benz, 2011). Many factors influencing the mass transfer on bioprocess such as: temperature, pressure, concentration, diffusivity, viscosity, density, pH, fluid flow rate, bioreactor geometry, surface tension (Monit, 2009).

### 1.2. Anaerobic Bioreactor

Anaerobic treatment principally utilizes the anaerobic bacteria in order to degrade dissolved organic materials or soluble chemical oxygen demand (SCOD) to become biogas (Abdelgadir et al., 2014). The process of anaerobic decomposition degrades natural polymer, such as polysaccharides, proteins, nucleic acids, and lipids into methane and carbon dioxide. It occurs in gradual and parallel reaction. The efficiency of anaerobic bioreactor treatment is sensitive towards the composition of wastewater, temperature, and pH (Merlin et al., 2012).

Anaerobic treatment of organic material is a complex and specific biochemical reaction. Biodegradation of dissolved organic material undergoes reaction stages: hydrolysis, acidogenesis, acetogenesis and methanogenesis that happen simultaneously, either in serial or parallel ways (Zinatizadeh et al., 2006; van Lier et al., 2008; Deublein and Steinhauser, 2008; Appels et al., 2008; Abdelgadir et al., 2014). Bacteria that plays the role on the four stages works specifically and has mutual dependence (Hossain et al., 2009).

Hydrolysis is a liquifaction of organic materials using extra-celular enzyme produced by hydrolytic bacteria (Zinatizadeh et al., 2006; Deublein and Steinhauser, 2008; Appels et al., 2008). Celulolytic bacteria holds the role in hydrolysis stage, works in the pH rate of 6-7. In this process, the likelihood of pH degradation may happen due to the formation of organic acids. Hydrolysis depends on the parameter such as: particle size, pH, enzyme production, diffusion and adsorption of enzyme on the waste particle that undergoes decomposition process (Ziemiński and Frac, 2012), the substrate concentration and temperature (Gerber and Roland, 2008). Another research reported that the rate of hydrolysis constant depends on pH but not coefficient with the total and concentration of VFA (Veeken et al., 2000). And the best pH range to reach maximum biogas result in anaerobic bioreactor is 6.5-7.5 (Liu et al., 2008; Khalid et al., 2011) and 6.8-7.2 (Ward et al., 2008).

Acidogenesis is a step of the recast of organic materials resulted from hydrolysis to become amino acid, simple

sugar and VFA, including format acid, acetic, propionic, butyric, lactic, succinic, ethanol, and CO<sub>2</sub>, H<sub>2</sub>, NH<sub>3</sub>, H<sub>2</sub>S by acid-forming bacteria (van Lier et al., 2008; Ziemiński and Frac, 2012). For batch system experiment, the range of pH on stage thermophilic acidogenesis is 6-7 (Park et al., 2008). VFA is an intermediate product which is very important in anaerobic treatment (Appels et al., 2008). Acetogenesis is a stage of forming acetic compound, carbondioxyde and hydrogen (Ziemiński and Frac, 2012). It is reported that methanogenesis in anaerobic bioreactor is efficient on the pH of 6.5-8.2 (Lee et al., 2009), while hydrolysis and acidogenesis respectively on the pH of 5.5 and 6.5 (Kim et al., 2009). The last step in anaerobic biodegradation is methanogenesis. Most methanogens bacteria is mesophilic with the range of temperature between 28-42°C and thermophilic temperature with the range of 55-72°C (Ziemiński and Frac, 2012). Optimum methanogens bacteria on the pH range of 7.2-8.0 (Suryawanshi et al., 2013).

### 1.3. Mass Transfer and Diffusion of Dissolved Material

The coefficient of liquid-phase mass transfer is a function of liquid physical characteristic and the liquid superficial velocity (Zaiat et al., 2000; Abdelgadir et al., 2014). If the liquid-phase mass transfer becomes the limitation of the whole process rate, then  $k_L$  can be estimated with Equation 1 (Zaiat et al., 2000; Cho and Young, 2001).

$$N_L = k_L (S_b - S_i) \quad (1)$$

Where  $N_L$  is the flux of liquid-phase mass transfer, (mg/cm<sup>2</sup>.h);  $k_L$  is the coefficient of liquid-phase mass transfer;  $S_b$  is the substrate concentration on the bulk liquid, (mg COD/L); and  $S_i$  is the substrate concentration on the solid-liquid interface, (mg COD/L).

Mass transfer depends on the diffusion of dissolved materials and the variables controlling the character of fluid flow such as; flow rate, viscosity, density, and a linear dimension. Diffusion based on empirical modification reported by Wilke-Chang (Geankoplis, 2003), where the solution diffusion can be calculated using Equation 2.

$$D_L = 7,4 \times 10^{-8} \frac{(\psi_B \cdot M_B)^{1/2} T}{\mu_L \cdot V_A^{0,6}} \quad (2)$$

Where,  $D_L$  is liquid-phase diffusion of the dissolved materials and the solvent, cm<sup>2</sup>/s;  $T$  is temperature, K;  $\mu_L$  is viscosity of the solution, cP;  $V_A$  is the molar volume of the dissolved material in normal boiled point, cm<sup>3</sup>/gmol;  $\psi_B$  is the parameter of solvent association (water = 2.6) and  $M_B$  is the weight of solution molecule (18 g/gmol).

Several studies reported that bioreactor failures or deficient performance due to pH degradation is caused by the accumulation VFA within the system of anaerobic treatment (Visser et al., 1993; Fabián dan Gordon, 1999; Poh and Chong, 2009; Thabathaba'i et al., 2011). The increase of VFA concentration in batch system bioreactor

gives different influence towards hydrolysis, acidogenesis, and methanogenesis stages (Appels et al., 2008). The high VFA concentration will obstruct hydrolysis, acidogenesis, and methanogenesis that the adequate process of mass transfer and microorganism seeding become important things in anaerobic bioreactor (Vabilin et al., 2002).

The objectives of this research is to decide the pH influence towards the coefficient of liquid-phase mass transfer;  $k_L$ , and the diffusion of dissolved material,  $D_L$  on the leachate treatment in anaerobic bioreactor.

## 2. Method

Leachate used was from Sambutan Landfill, Samarinda, East Borneo, Indonesia. The anaerobic bioreactor used was featured with heater, leachate recirculation pump, leachate recirculation flowmeter, manometer, leachate entrance inlet, biogas thermometer, pressure gauge, leachate thermometer, sampling port, leachate recirculation entrance faucet, leachate recirculation exit faucet and leachate effluent faucet. The type of this research is pilot scale experiment with semi-batch system.

This research uses anaerobic bioreactor with the total volume  $\pm$  160 L. After doing characterization and analysis of leachate quality, subsequently the designing of anaerobic bioreactor with the design according to the need of the research was done. Then the test of leakage and calibration from the system of anaerobic bioreactor was done.

Seeding and acclimatization were done, respectively for 10 days. Then it was continued by anaerobic treatment for 21 days. Seeding and acclimatization were done in the anaerobic bioreactor on the ambient temperature with pH variation: ambient, 7.2 and 8.0, where the ratio of leachate volume: biogas was 70:30. Microorganism used was from cow rumen and leachate with the ratio of 1:3 and filtered for the extract. Analysis and parameter test of COD and VFA were done once in two days. The process of leachate treatment was stopped if the decreasing percentage of COD (COD removal) reached 60-80%.

## 3. Result and Discussion

### 3.1. Results

The condition of seeding and acclimatization was done on pH ambient. On seeding, the leachate temperature started from 27.4-29 °C, biogas temperature of 26-27.5 °C, pH decreased from 7.5-7.3, and the biogas pressure started to increase on the 9<sup>th</sup> day, for 1-1.5 mm H<sub>2</sub>O. Meanwhile on acclimatization, the leachate temperature started for 28.2-28.5 °C, biogas temperature of 27-27.9 °C, pH increased from 7.2-7.7, and biogas pressure increased from 4-21 mm H<sub>2</sub>O, with the average value of BOD/COD 0.58.

The condition of seeding and acclimatization was done on pH of 7.2. On seeding, the leachate temperature started from 28.2-29 °C, biogas temperature of 27.5-28.5 °C, and the biogas pressure started to increase on the 6<sup>th</sup> day, with the up and down fluctuation between 1 up to 5 mm H<sub>2</sub>O. Meanwhile on acclimatization, the leachate temperature started for 27-27.5 °C, biogas temperature of 26-27.5 °C,

and biogas pressure increased from 7-29 mm H<sub>2</sub>O, with the average value of BOD/COD 0.70.

The condition of seeding and acclimatization was done on pH of 8.0. On seeding, the leachate temperature started from 27-28.5 °C, biogas temperature of 26-27.9 °C, and the biogas pressure started to increase on the 8<sup>th</sup> day, with the up and down fluctuation between 1-2 mm H<sub>2</sub>O. Meanwhile on acclimatization, the leachate temperature started for 27-28.5 °C, biogas temperature of 26-27.9 °C, and biogas pressure increased from 3-17 mm H<sub>2</sub>O, with the average value of BOD/COD 0.65.

### 3.2. Mass Transfer Rate, $r_{kL}$ and Diffusion Rate of Dissolved Material, $r_{DL}$

On pH ambient, the average mass transfer rate per day was the lowest of 0.00407 mg/L.day and the highest on the 27<sup>th</sup> day which was 0.14134 mg/L.day. The rate of dissolved material diffusion ranged between  $5.57442 \times 10^{-5}$  -  $1.40034 \times 10^{-4}$  cm<sup>2</sup>/s. On the pH of 7.2, the average rate of mass transfer per day was the lowest of 0.00227 mg/L.day and the highest on the 25<sup>th</sup> day was 0.16953 mg/L.day. The rate of dissolved material diffusion initially decreased and then increased, with the range between  $5.96009 \times 10^{-5}$  -  $1.44715 \times 10^{-4}$  cm<sup>2</sup>/s. Meanwhile on the leachate treatment within anaerobic bioreactor with the pH of 8.0, the average rate of mass transfer per day was the lowest of 0.02957 mg/L.day and the highest on the 37<sup>th</sup> day was 0.13090 mg/L.day. The rate of dissolved material diffusion went up and down, with the range between  $5.54146 \times 10^{-5}$  -  $1.2416 \times 10^{-4}$  cm<sup>2</sup>/s.

A research about pH influence on two-phased anaerobic bioreactor reported that acidogenic bioreactor preserved on the pH of 6.0-7.0; resulted in the decrease of SCOD from 6000 mg/L to 1000-1500 mg/L, that the efficiency of COD removal increased from 50% to 80%. And the methane production of 0.32 m<sup>3</sup> CH<sub>4</sub>/kg COD removed with methane contents on the methanogenic bioreactor of 80-90%. Meanwhile, without the pH control, the pH condition on acidogenic bioreactor increased up to 8.2; the acid conversion decreased yet the COD degradation was almost similar to the pH controlled on 6.0-7.0. Only without the pH control, methane contents from methanogenic bioreactor increases more for 90% (Jung et al., 2000).

### 3.3. Discussion

#### 3.3.1. pH

pH in seeding and acclimatization steps, went up and down from the lowest of 6.3 and increased up to 7.9. As seen in Figure 1, where seeding and acclimatization steps were done on the first day until the 20<sup>th</sup> day. It can be seen that seeding step on pH ambient ranged between 6.2 up to 7.8. Meanwhile, acclimatization step on pH ambient ranged between 6.3 up to 8.0.

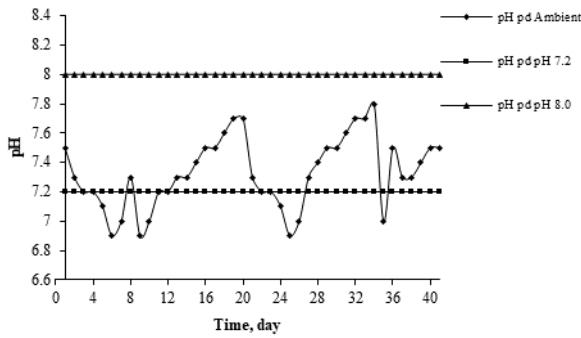


Figure 1. pH on multiple variations

pH is one of important parameters on anaerobic treatment because methanogenic bacteria is very sensitive towards the change of pH. Methane-forming bacteria live well on neutral to a little alkali condition. pH in bioreactor is directly dependent on the retention time (Kigozi et al., 2014).

pH on both stages, seeding-acclimatization steps and anaerobic treatment, showed the difference of pH range. Where pH of seeding-acclimatization was on a little acid pH while pH of anaerobic treatment was on a little alkali pH. It occurred due to the fact that biodegradation stage involved hydrolysis bacteria which produced extra-cellular enzyme (Abdelgadir et al., 2014). This group of cellulolytic bacteria optimally played the role on the pH ranging between 6-7. Dihydrolysis protein became amino acid by the protease functioning as exo-enzyme. Hydrolysis lipid became glycerol and fat acid. Hydrolysis hydrocarbon became glucose and other simple sugars. Hydrolysis was done by facultative anaerobic bacteria through exo-enzyme (Wiesmann et al., 2007). In this process, the likelihood of pH decrease may happen due to the formation of organic acids.

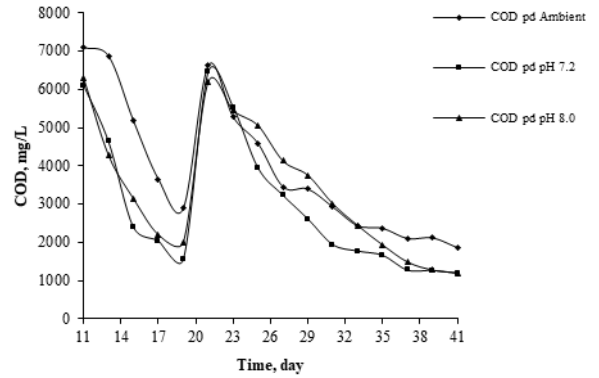
In initial stage, a large number of VFA is produced by acid-forming bacteria, pH in bioreactor may decrease up to below 5. This condition can obstruct or stop the process within bioreactor. Methanogenic bacteria is very sensitive towards the change of pH and cannot develop on the pH below 6.5 despite the fact that the process can still run within pH range between 6.0-8.0 (Kigozi et al., 2014).

Meanwhile, on anaerobic treatment stage, where anaerobic biodegradation that played the role was methanogenesis bacteria; hydrogenophilic or hydrogenotrophic, that formed methane from CO<sub>2</sub> and H<sub>2</sub> and methanogens acetoclastic or acetotrophic, resulting in methane by acetate decarbonization manner (Sekiguchi dan Kamagata, 2004). The group of methanogens bacteria utilized the result from three biogas. Methanogens bacteria is optimally mesophilic within the temperature range of 28-42 °C (Ziemiński and Frac, 2012) and optimum on the pH range of 7.2 – 8.0 (Suryawanshi et al., 2013).

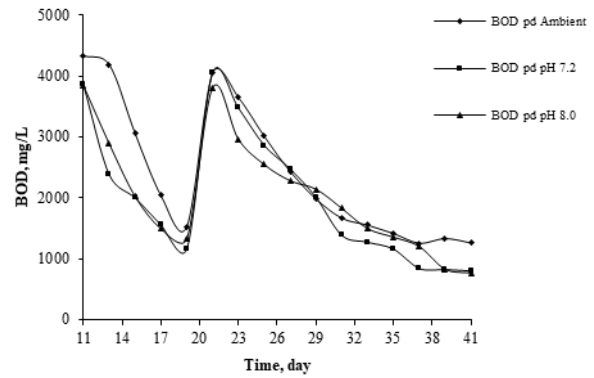
### 3.3.2. COD and BOD

The indicator of microorganism success in biodegradation is the decrease of substrate concentration (COD and BOD) in the leachate, as shown in Figure 2.

The substrate concentration (COD and BOD) on acclimatization stage, 11<sup>th</sup> day up to 21<sup>st</sup> day, decreased, as shown on figure 4.10 and 4.11. The minimal decrease of COD was 40.42% and maximal of 75.07%. Meanwhile the minimal decrease of BOD was 41.97% and maximal of 72.78%.



a.



b.

Figure 2. a. COD and b. BOD on pH variations

The decrease of COD and BOD on acclimatization stage was caused by the decrease of dissolved substrate concentration utilized by microorganism to grow, reproduce, and adapt within the leachate. Therefore, in this stage, the hydrolysis undissolved complex organic molecules become simple molecules dissolved in the leachate. It indicates that the growth of bacteria goes well. It can also be seen that the value of biodegradability ratio BOD/COD ranges between 0.43-0.70. The process of anaerobic treatment is effective enough for leachate with the high BOD:COD ratio resulted from the landfill initial stage (Li et al., 2010).

After the acclimatization stage entered the anaerobic treatment stage, leachate was added on the 21<sup>st</sup> day, COD and BOD increased again, as seen in figure 2. Such increase occurred due to the addition of organic and inorganic materials dissolved in the leachate within the bioreactor. Therefore, COD and BOD increased due to the addition of substrate concentration to the leachate.

The pH also affects the decrease of substrate concentration in anaerobic treatment stage, the 21<sup>st</sup> until 41<sup>st</sup> days. The percentage of COD decrease on pH ambient,

pH 7.2 and pH 8.0 respectively were 71.84%, 81.43% and 80.55%. Therefore, COD removal on pH 7.2 > pH 8.0 > pH ambient. Meanwhile BOD removal obtained ranged between 68.91% and 80.29%. With average COD and BOD removal respectively were 77.94% and 76.37%.

On overall stages, seeding-acclimatization and anaerobic treatment, the COD and BOD removal were on pH 7.2 > pH 8.0 > pH ambient. The bigger COD reduction was, the bigger the organic material degraded to organic acids. Organic acids were subsequently converted to methane gas. Therefore, the bigger reduction of COD was, the bigger the rate of methane gas formation was.

### 3.3.3. VFAs

VFA is an intermediate product (Appels et al., 2008) that will be converted on acetogenesis stage to become acetate acid (Gerardi, 2003; Ganidi et al., 2009), CO<sub>2</sub> and H<sub>2</sub> (Ziemiński and Frac, 2012). The VFA concentration is determined as a parameter in order to figure out how far the acidogenesis and acetogenesis stages go. Where VFA concentration becomes one of the good parameters to monitor in deciding the stability of anaerobic bioreactor.

The higher the pH is, the higher the obtained VFA concentration is. The VFA concentration is influenced by pH. On pH influence, it is obtained that the VFA concentration ranges between 83.33–370.63 mg/L. However, it still considers the optimum pH condition of the microorganism involved in such biodegradation. Where

VFA and pH ambient > pH 7.2 > pH 8.0 as seen in Figure 3.

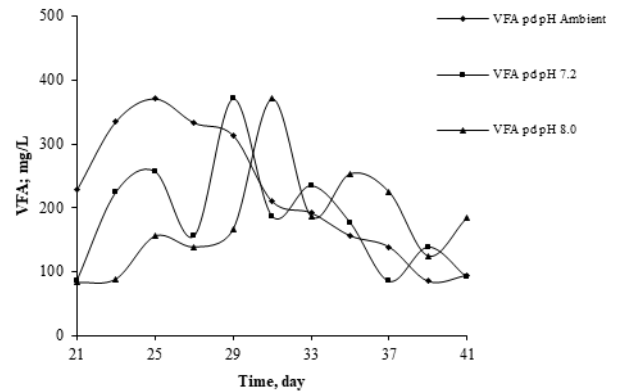


Figure 3. VFA from pH variation

As seen in figure 4, in the initial treatment, VFA increase is accompanied by the decrease of substrate concentration (COD and BOD). As time goes by, the bigger the COD and BOD removal are, the more decreasing VFA is. It is an undeniable fact that the exponential increase of VFA decreases again; it shows that the dependence of VFA towards the limited substrate concentration in the leachate (Reid et al., 1991). It can also be seen that the highest VFA concentration in all operation condition was on the 25<sup>th</sup> - 31<sup>st</sup> days.

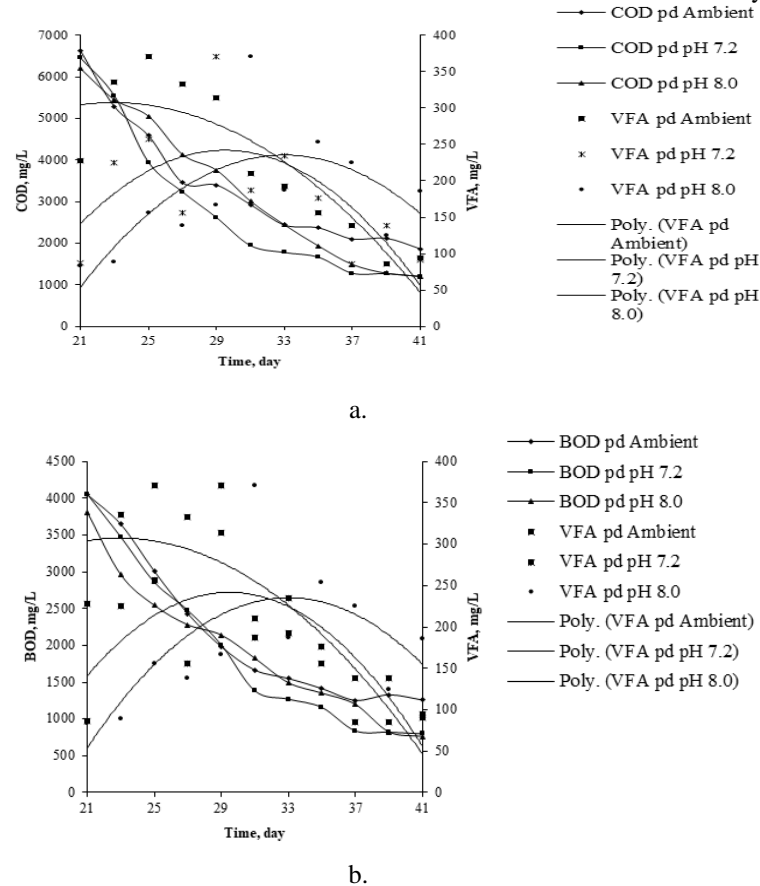


Figure 4. a. COD vs VFA and b. BOD vs. VFA

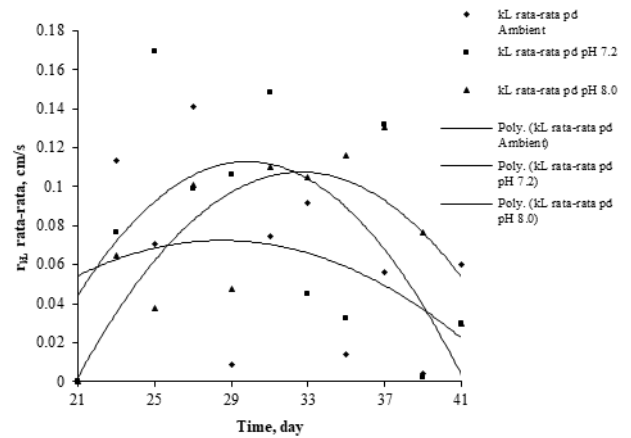
The VFA accumulation within anaerobic bioreactor represented imbalance between acid produced and consumed by bacteria. If the bioreactor overloads and the content of VFA concentration is high, bigger than what the methane-producing bacteria consumes (slow-growers), then the biogas production will increase. This increase potentially improves the foaming in anaerobic bioreactor (Baber, 2005). The higher the biodegraded COD is, the higher VFA concentration is. The more decreasing the substrate concentration is, the more decreasing the VFA concentration is. It is in line with the statement that the higher reduced substrate concentration is, the bigger the biodegraded dissolved organic material to become organic acids is. This VFA is subsequently converted into methane gas. Therefore, if the decrease of COD gets bigger, then the rate of biogas formation becomes higher.

**3.3.4. Liquid-Phase Mass Transfer and Diffusion of Dissolved Material**

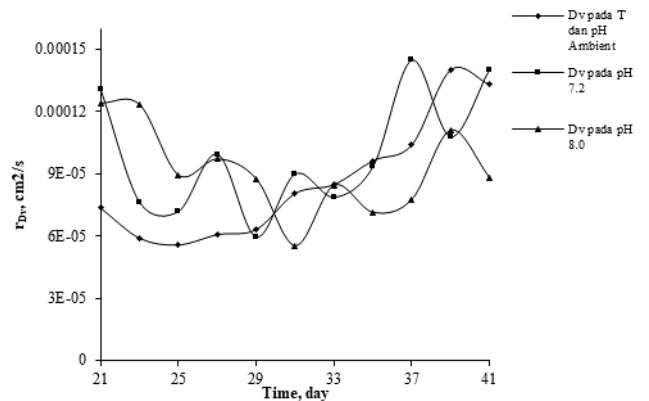
The mass transfer and dissolved material diffusion in the leachate also depends on the change of dissolved substrate concentration within the leachate. The average of mass transfer per day is the number dissolved material substrate biodegraded in each day. COD degradation is accompanied by the increase of mass transfer rate, average  $r_{kL}$  per day. Diffusion of dissolved material,  $D_L$  decreases by the increase of the pressure. As seen in figure 5, figure 6, and figure 7.

On pH ambient, it is obtained that the average lowest  $r_{kL}$  per day is 0.00407 mg/L.day and the highest on the 27<sup>th</sup> day is 0.14134 mg/L.day.  $r_{DL}$  of the dissolved material ranges between  $5.57442 \times 10^{-5}$  -  $1.40034 \times 10^{-4}$  cm<sup>2</sup>/s. On pH of 7.2, the average lowest  $r_{kL}$  per day is 0.00227 mg/L.day and the highest on the 25<sup>th</sup> day is 0.16953 mg/L.day. The coefficient of liquid-phase mass transfer is 0.08400 cm/s.  $r_{DL}$  of the dissolved material initially decreases and then increases, with the range between  $5.96009 \times 10^{-5}$  -  $1.44715 \times 10^{-4}$  cm<sup>2</sup>/s. Meanwhile with the pH of 8.0, the obtained lowest average  $r_{kL}$  per day is 0.02958 mg/L.day and the highest on the 37<sup>th</sup> day is 0.13090 mg/L.day. The

coefficient of liquid-phase mass transfer is 0.09200 cm/s.  $r_{DL}$  of the dissolved material initially decreases and then increases, with the range between  $5.54146 \times 10^{-5}$  -  $1.2416 \times 10^{-4}$  cm<sup>2</sup>/s.

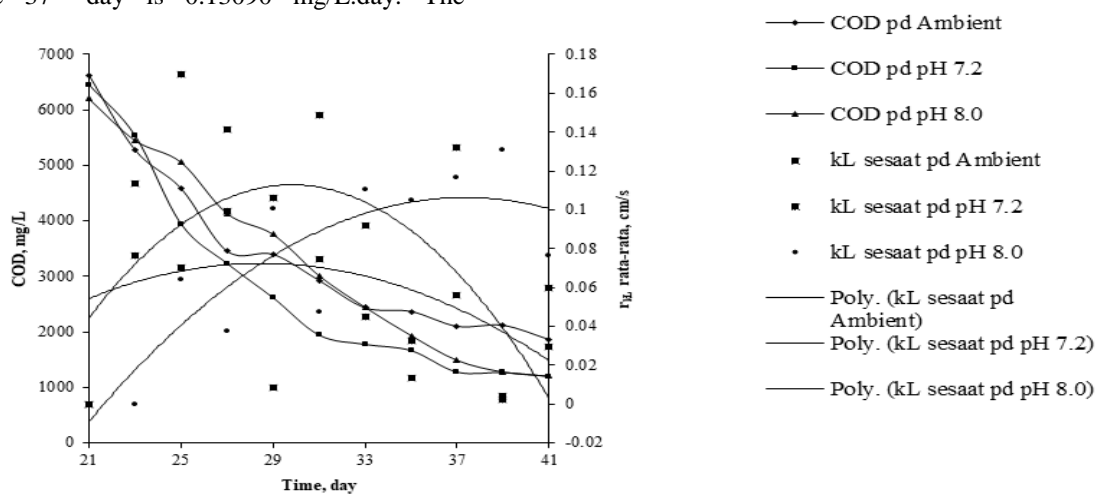


a.

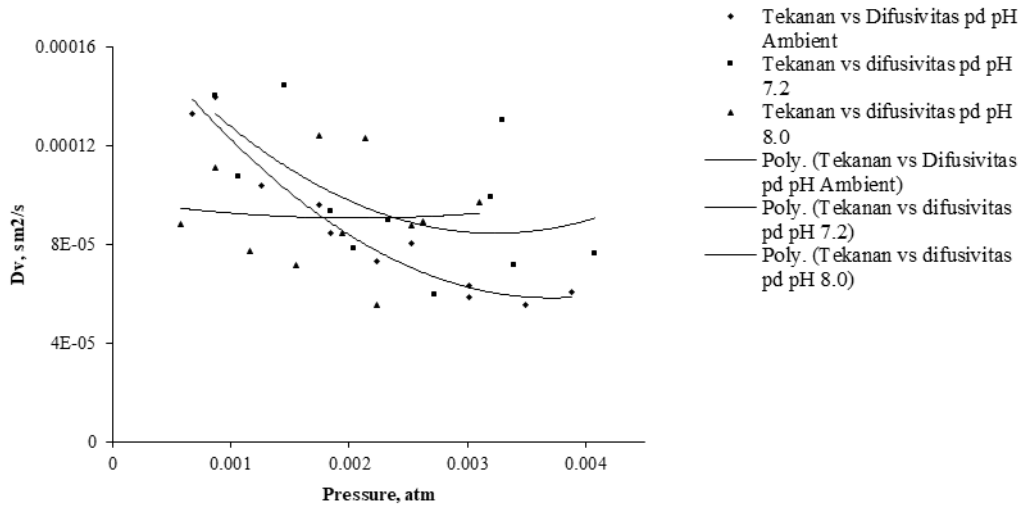


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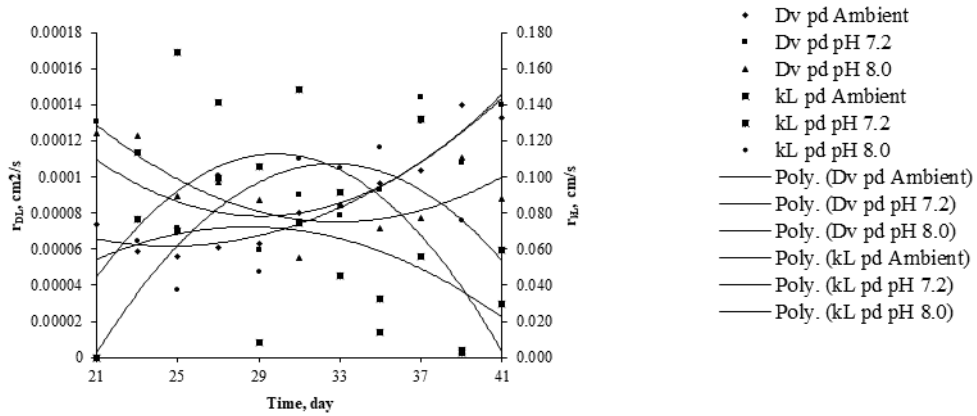
**Figure 5.  $r_{kL}$  and  $r_{DL}$  on pH variations**



a.



b. **Figure 6.** a. COD vs  $r_{KL}$  on pH variation; b. Diffusivity vs Biogas Pressure on pH variation



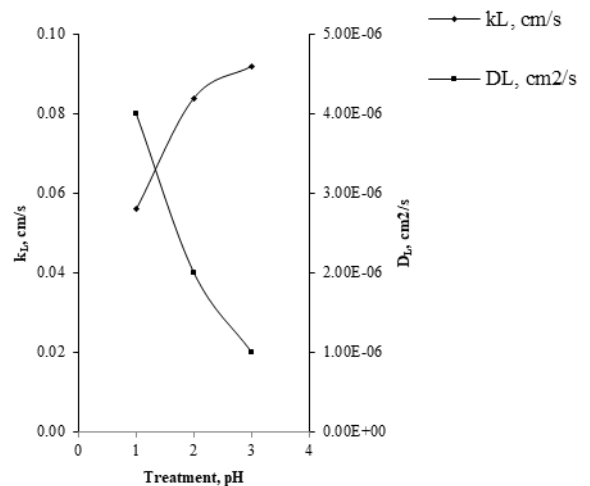
**Figure 7.**  $r_{DL}$  vs  $r_{KL}$  on pH variation

The  $k_L$  and  $D_L$  of leachate treatment within the anaerobic bioreactor done on pH ambient, pH 7.2 and pH 8.0. Where the coefficient of liquid-phase mass transfer increases from 0.056 - 0.092 cm/s, while the diffusivity of dissolved materials decreases from  $4.10 \times 10^{-6}$  to  $1.10 \times 10^{-6}$  cm<sup>2</sup>/s, as seen from Table 1.

**Table 1.** Liquid-phase Mass Transfer Coefficient,  $k_L$ , and Coefficient of Dissolved Materials Diffusion;  $D_L$

Treatment, pH	$k_L$ , cm/s	$D_L$ , cm <sup>2</sup> /s
Ambient	0.056	$4.10^{-6}$
7.2	0.084	$2.10^{-6}$
8.0	0.092	$1.10^{-6}$

On the treatment of pH variation, it can be seen that the higher pH is, the more increasing  $k_L$  is and the more increasing the  $D_L$  is.  $k_L$  pH 8.0 >  $k_L$  pH 7.2 >  $k_L$  pH ambient. Meanwhile  $D_L$  pH ambient >  $D_L$  pH 7.2 >  $D_L$  pH 8.0, as seen in Figure 8.



**Figure 8.**  $D_L$  vs  $k_L$  on pH variations

**Table 2.** Research on Mass Transfer

Bioreactor	Wastewater	V <sub>R</sub> , L	HRT(days)	Temp. (°C)	COD <sub>Inf.</sub> (mg/L)	COD <sub>Rem.</sub> (%)	Mass Transfer Coefficient, <i>k<sub>L</sub></i>	References
HAIB (pilot-scale)	Glucose-based synthetic substrate	237	8 h	30	2.090-41	98	3.40.10 <sup>-2</sup> cm/h	Zaiat et al., 2000
	Domestic Sewage		8 h	30	341-71	79.2	2.23.10 <sup>-1</sup> cm/h	Zaiat et al., 2000
Two-stage Anaerobic Filter	Brewery wastewater	7.45 and 6.55	0.5-6 d	35	1500-2500 (OLR 0.5-20 g SCOD/L.d)	98.2	1.4-2.2 d <sup>-1</sup>	Cho and Young, 2001
Fixed-Bed ASBR	Synthetic Wastewater (Polyurethane Foam)	1.2	8 h	30	500-68	72-87	1.98-1.85 h <sup>-1</sup>	Ramos et al., 2003
UASB	Synthetic Wastewater (phenol)	3.78	0.97-1.03 d	35	10.53-12.61 kg COD/m <sup>3</sup> .d	99.3-96.3	19.2 mg phenol/L	Chou and Huang, 2005
SASBR	Synthetic Wastewater (Polyurethane Foam)	5		30	285-333	84-89	0.48-0.60 h <sup>-1</sup>	Cubas et al., 2007
Anaerobic Batch Bioreactor (pilot-scale)	Leachate	160	1 d	Ambient (27-30)	6200-6625.4	71.84-81.43	0.056, 0.084 and 0.092 cm/s	This research

Annotation:

HAIB : Horizontal-Flow Anaerobic Immobilized Biomass

UASB : Upflow Anaerobic Sludge Bed Reactors

ASBR : anaerobic sequencing batch Reactor

SASBR : Stirred Anaerobic Sequencing Batch Reactor

The large number of wastewater used in the research on mass transfer is synthetic wastewater (Zaiat et al., 2000; Ramos et al., 2003; Chou and Huang, 2005; Cubas et al., 2007) and on laboratory scale (Cho and Young, 2001; Ramos et al., 2003; Chou and Huang, 2005; Cubas et al., 2007). The research done by Souza et al. (2014) aims at finding out the density and viscosity of leachate on different temperatures. Meanwhile, the research done by Vavilin et al. (2002), on anaerobic bioreactor of solid waste on the influence of leachate recirculation and pH aims at analyzing the balance between hydrolysis rate/organic polymer acidogenesis and methanogenesis. The comparison of the research done in this research is shown in Table 2.

#### 4. Conclusion

The pH affects the decrease of concentration of dissolved organic substrate (COD and BOD) of the leachate. The higher the pH is, the higher the obtained VFA concentration is. VFA concentration is influenced by pH, but still considering the optimum pH condition of the biodegraded substrate. pH affects the average mass transfer rate, *r<sub>KL</sub>* on the anaerobic treatment of leachate. pH affects the concentration of dissolved organic substrate which subsequently influences the coefficient of liquid-phase mass transfer of leachate, *k<sub>L</sub>* and diffusivity of dissolved material, *D<sub>L</sub>*. The higher the pH is, the more increasing the *k<sub>L</sub>* is and the more decreasing the *D<sub>L</sub>* is. The *k<sub>L</sub>* pH 8.0 > *k<sub>L</sub>*

pH 7.2 > *k<sub>L</sub>* pH ambient. Meanwhile, *D<sub>L</sub>* pH ambient > *D<sub>L</sub>* pH 7.2 > *D<sub>L</sub>* pH 8.0. The coefficient of liquid-phase mass transfer increases from 0.056–0.092 cm/s, while the diffusivity of dissolved material decreases from 4.10 x 10<sup>-6</sup> to 1.10 x 10<sup>-6</sup> cm<sup>2</sup>/s.

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#### Nomenclature

*N<sub>L</sub>* is flux of liquid-phase mass transfer, (mg/cm<sup>2</sup>.h)

*k<sub>L</sub>* is coefficient of liquid-phase mass transfer

*S<sub>b</sub>* is substrate concentration on bulk liquid, (mg COD/L)

*S<sub>i</sub>* is substrate concentration on solid-liquid interface, (mg COD/L)

*D<sub>L</sub>* is liquid-phase diffusion of dissolved material and solvent, (cm<sup>2</sup>/s)

*T* is temperature, (K)

*μ<sub>L</sub>* is viscosity of solvent, (cP)

*V<sub>A</sub>* is molar volume of dissolved material in normal boiling point, (cm<sup>3</sup>/gmol)

*Ψ<sub>B</sub>* is the parameter of solvent association (air = 2.6)

*M<sub>B</sub>* is the weight of solvent molecule (18 g/gmol)



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